## Total Synthesis of ( $\pm$ )-Lubimin and ( $\pm$ )-Oxylubimin

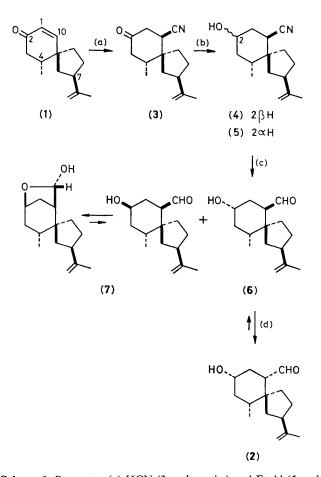
Akio Murai, Shingo Sato, and Tadashi Masamune\*

Department of Chemistry, Faculty of Science, Hokkaido University, Sapporo 060, Japan

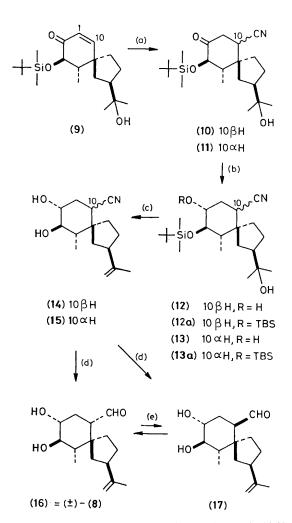
The total synthesis of the title compounds, examples of the spirovetivane type of phytoalexins in the genus *Solanum*, by transformation of  $(\pm)$ -15-norsolavetivone and its derivatives, is described.

The title compounds, described as phytoalexins, are members of a group of sesquiterpenes of the spirovetivane type, produced by plants of the *Solanum* genus.<sup>1,2</sup> These stress metabolites are biosynthetic intermediates in the major pathway from acetic acid to rishitin *via* solavetivone *in vivo*,<sup>3,4</sup> and are characterized structurally by the presence of five or six asymmetric centres in the spiro[4.5]decane system. In the preceding communication<sup>5</sup> we reported the total synthesis of  $(\pm)$ -15norsolavetivone and related compounds. We describe herein transformation of these compounds into  $(\pm)$ -lubimin,  $(\pm)$ oxylubimin, and related compounds, which constitutes the first synthesis of these highly oxygenated spirovetivane phytoalexins.

Conversion of  $(\pm)$ -15-norsolavetivone (1) into  $(\pm)$ lubimin (2) (Scheme 1) commenced with hydrocyanation of (1) by a modification of Nagata's conditions.<sup>6</sup> The reaction proceeded regio- and stereo-selectively, giving 10-cyano-15norsolavetivone (3) as the sole product (81%). The n.m.r.



spectrum [ $\delta$  2.61 (2H, d, J 6 Hz, 1-H) and 3.10 (1H, t, J 6 Hz, 10-H)] indicated that (3) is a mixture of rapidly equilibrating conformers. Reduction of (3) with borane-ammonia complex<sup>7</sup> gave a mixture of 2-eq- (4) and 2-ax- (5) hydroxy-10-ax-cyano-15-norsolavetivones, which were easily separated by chromatography in 60 and 30% yields, respectively [(4),  $\delta$  2.74 (1H, t, J 4 Hz, 10-H) and 3.97 (1H, br. m,  $W_{\frac{1}{2}}$  25 Hz, 2-H); (5),  $\delta$  2.77 (1H, t, J 4 Hz) and 3.92 (1H, br. m,  $W_{\frac{1}{2}}$  12 Hz)]. Compounds (4) and (5), when treated with disobutylaluminium hydride in ether, were converted into 2-eq-hydroxy-10-ax-formyl-15-norsolavetivene (6) and its 2-ax-epimer (7) in 93 and 74% yields, respectively [(6), i.r., 1715 cm<sup>-1</sup>,  $\delta$  9.85 (1H, s, 15-H); (7) (isolated as a 1:3 insepar-



Scheme 1. Reagents: (a) HCN (3 mol. equiv.) and  $Et_aAl$  (5 mol. equiv.) in tetrahydrofuran, room temp., 4 h. (b)  $BH_a$ - $NH_a$  complex in aqueous MeOH, room temp., 2 h. (c) Di-isobutyl-aluminium hydride in ether, 0 °C, 3 h. (d) 5% KOH in MeOH, room temp., 1 h.

Scheme 2. Reagents: (a) HCN (3 mol. equiv.) and  $Et_3Al$  (5 mol. equiv.) in tetrahydrofuran, room temp., 2 h. (b)  $BH_3-NH_3$  complex in aqueous MeOH, room temp., 3.5 h; TBSCl-imidazole in dimethylformamide, 50 °C, 36 h. (c) Pyridine $-Al_2O_3$ , 220 °C, 30 min; HF in tetrahydrofuran-MeCN, room temp., 6.5 h, (d) Di-isobutylaluminium hydride in dimethoxyethane, room temp., 12 h. (e) 5% KOH in MeOH, room temp., 1 h. TBS =  $Bu^{\dagger}Me_2Si$ .

able mixture of the hydroxy aldehyde and its acetal), i.r., 3630, 3450, 2730, and 1719 cm<sup>-1</sup>,  $\delta$  4.02 (0.25H, m,  $W_{\frac{1}{2}}$  10 Hz, 2-H), 4.33 (0.75H, m,  $W_{\frac{1}{2}}$  16 Hz, 2-H), 5.18 (0.75H, br. s, 10-H), and 9.73 (0.25H, m,  $W_{\frac{1}{2}}$  6 Hz, 10-H)]. Compounds (6) and (7) were identified as  $(\pm)$ -10-epilubimin<sup>8</sup> and  $(\pm)$ -2-epi-10-epilubimin<sup>9</sup> by direct comparison with natural samples. After repeated epimerization of (6),<sup>8</sup>  $(\pm)$ -lubimin (2)<sup>2</sup> was isolated in an overall yield of 37% from  $(\pm)$ - (1) (2.1% from orcinol dimethyl ether<sup>5</sup>).

The synthesis of  $(\pm)$ -oxylubimin (8) was performed in the same manner as that of  $(\pm)$ -(2), starting with the 3-t-butyldimethylsilyl ether (9) of (4RS, 7SR)-3,11-dihydroxy-15norspirovetiv-1(10)-en-2-one<sup>5</sup> (Scheme 2). Hydrocyanation of (9) led to only regioselective formation of a mixture of the corresponding, easily separable 10-eq-(10) (44%) and 10-ax-(11) (32%) cyano-norsolavetivanes [(10),  $\delta$  2.88 (1H, dd, J 12) and 4.5 Hz, 10-H) and 3.77 (1H, d, J 11 Hz, 3-H); (11), 8 3.05 (1H, t, J 5 Hz) and 3.76 (1H, d, J 11 Hz)]. Reduction of (10) and (11) with borane-ammonia complex proceeded stereoselectively, in contrast with (3), giving the corresponding 2alcohols (12) and (13), which on silvlation formed the respective 2,3-disilyl ethers (12a) and (13a) in 80 and 58% yields [(12a),  $\delta$  3.26 (1H, t, J 7 Hz, 3-H) and 3.50 (1H, m,  $W_{\frac{1}{2}}$ 20 Hz, 2-H); (13a), § 3.52 (1H, t, J 4 Hz) and 3.72 (1H, m,  $W_{4}$  12 Hz)]. Dehydration of the isopropyl moieties of (12a) and (13a) by treatment with pyridine-modified alumina<sup>10</sup> followed by removal of the silvl groups<sup>11</sup> afforded 10-eq- (14) and 10-ax-(15) cyano-2,3-di-eq-hydroxy-15-norsolavetivenes in 67 and 47% yields, respectively. Treatment of (14) and (15) with di-isobutylaluminium hydride in dimethoxyethane gave the corresponding aldehydes (16), m.p. 79–81 °C, and (17), m.p. 121-123 °C, in 58 and 68 % yields, which were identified as  $(\pm)$ -oxylubimin<sup>2</sup> and  $(\pm)$ -10-epioxylubimin,<sup>8b,12</sup> respectively, by direct comparison with natural samples. After repeated epimerization<sup>8b,12</sup> of (17), ( $\pm$ )-oxylubimin was isolated in an overall yield of 20% from (9) (0.82\% from orcinol dimethyl ether<sup>5</sup>).

## Received, 23rd December 1981; Com. 1466

## References

- T. Masamune, A. Murai, and N. Katsui, Kagaku To Seibutsu, 1978, 16, 648; J. B. Stothers, Pure Appl. Chem., 1981, 53, 1241.
   N. Katsui, A. Matsunaga, H. Kitahara, F. Yagihashi, A.
- 2 N. Katsui, A. Matsunaga, H. Kitahara, F. Yagihashi, A. Murai, T. Masamune, and N. Sato, *Bull. Chem. Soc. Jpn.*, 1977, 50, 1217, and references cited therein.
- 3 K. Sato, Y. Ishiguri, N. Doke, K. Tomiyama, F. Yagihashi, A. Murai, N. Katsui, and T. Masamune, *Phytochemistry*, 1978, **17**, 1901; A. Murai, N. Katsui, F. Yagihashi, T. Masamune, Y. Ishiguri, and K. Tomiyama, *J. Chem. Soc.*, *Chem. Commun.*, 1977, 670.
- 4 A. Murai, S. Sato, A. Osada, N. Katsui, and T. Masamune, J. Chem. Soc., Chem. Commun., 1982, 32.
- 5 A. Murai, S. Sato, and T. Masamune, J. Chem. Soc., Chem. Commun. preceding communication.
- 6 (a) W. Nagata and N. Yoshioka, Org. React., 1977, 25, 255.
  (b) Addition of 2-lithio-1,3-dithian to (1) resulted in formation of a 1:1 mixture of 1,2- and 1,4-adducts: cf. C. A. Brown and A. Yamaichi, J. Chem. Soc., Chem. Commun., 1979, 100; M. El-Bouz and L. Wartski, Tetrahedron Lett., 1980, 21, 2897.
- 7 G. C. Andrews and T. C. Crawford, *Tetrahedron Lett.*, 1980, **21**, 693.
- 8 (a) N. Katsui, F. Yagihashi, A. Matsunaga, K. Orito, A. Murai, and T. Masamune, *Chem. Lett.*, 1977, 723. (b) N. Katsui, F. Yagihashi, A. Murai, and T. Masamune, *Bull. Chem. Soc. Jpn.*, in the press.
- 9 N. Katsui, Y. Takahashi, N. Sato, A. Murai, and T. Masamune, Nippon Kagaku Kaishi, 1981, 659.
- 10 Cf., E. von Rudloff, Can. J. Chem., 1961, 39, 1860.
- 11 R. F. Newton, D. P. Reynolds, M. A. W. Finch, D. R. Kelly, and S. M. Roberts, *Tetrahedron Lett.*, 1979, 3981, and references cited therein.
- 12 N. Katsui, F. Yagihashi, A. Murai, and T. Masamune, Chem. Lett., 1978, 1205.